

Chemistry 11 Syllabus

Textbook: Chemistry 4th ed. by Zumdahl

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| <p>Liquids and Solids (Ch. 10) Intermolecular forces Properties of liquids Types of solids Crystal lattices Metals Changes of state Phase diagrams</p> | <p>Chemical Kinetics (Ch. 12) Reaction rates Determining rate laws Reaction mechanisms Model for kinetics Catalysis</p> |
| <p>Properties of Solutions (Ch. 11) Solution composition Enthalpy of solution formation Factors affecting solubility Vapor pressure of solutions Colligative properties</p> | <p>Electrochemistry (Ch. 17) Galvanic cells Standard reduction potentials Cell potential and free energy Nernst equation Batteries Corrosion Electrolysis</p> |
| <p>Thermodynamics (Ch. 16) Entropy 2nd Law of Thermodynamics Free Energy Equilibrium</p> | <p>Organic Chemistry (Ch. 22) Nomenclature Functional groups Stereochemistry Reactions</p> |

Other resources

Schaum's Outline of College Chemistry by Rosenberg and Epstein. Contains elementary to advanced problems

Harcourt College Outline Series: College Chemistry by Goldwhite and Spielman. Contains mostly elementary and intermediate problems.

Both study guides can be found at any decent bookstore for about \$20.